

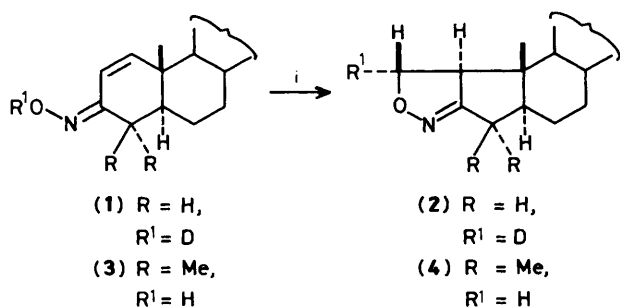
## A New Photoinduced Rearrangement of Steroidal $\alpha,\beta$ -Unsaturated Cyclic Ketone Oximes into the $\beta,\gamma$ -Unsaturated Isomer involving an Intramolecular Stereospecific Hydrogen Transfer<sup>1</sup>

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Irradiation of (*E*)-1,4,4-trimethyl-5 $\alpha$ -cholest-1-en-3-one oxime in a protic or an aprotic solvent gave rise to 4,4-dimethyl-1-methylene-5 $\alpha$ -cholestan-3-one oxime in 69–75% yield; mechanism for this photo-isomerization which involves an intramolecular stereospecific transfer of a hydroxyimino hydrogen is proposed on the basis of deuterium labelling studies.

We have recently reported<sup>2</sup> a new stereospecific rearrangement of an excited steroidal  $\alpha,\beta$ -unsaturated cyclic ketone oxime; irradiation of (*E*)-5 $\alpha$ -cholest-1-en-3-one oxime (**1**) in a protic or an aprotic solvent with a low pressure mercury arc



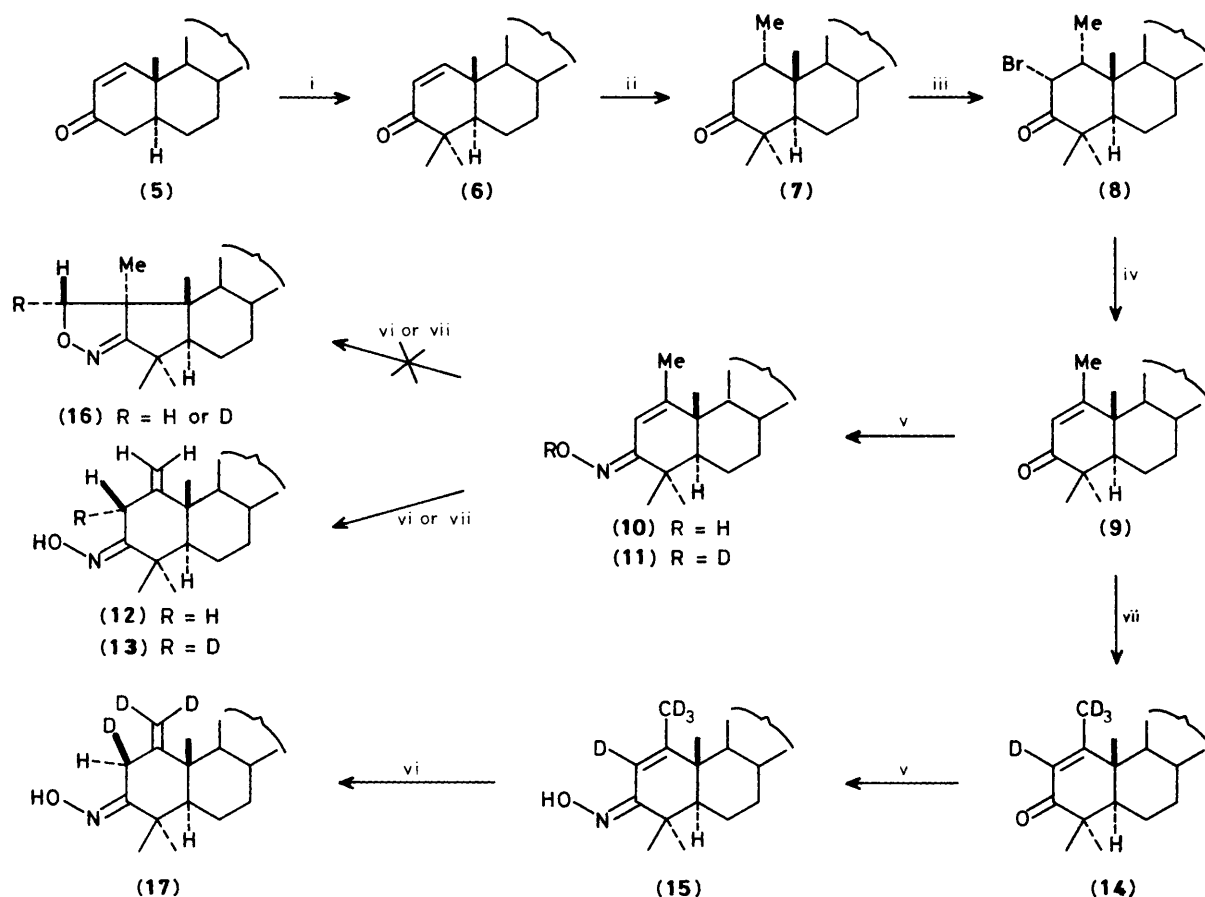
Scheme 1. Reagents and conditions: i, CD<sub>3</sub>OD or C<sub>6</sub>D<sub>6</sub>-h $\nu$ .

gave 4 $\alpha'$ ,5'-dihydro-*A*-nor-5 $\alpha$ -cholestano[2,1-*c*]isoxazole (**2**) (Scheme 1). On the basis of deuterium labelling studies, we advanced the hypothesis that the pathway involves an unusual stereospecific intramolecular transfer of the hydroxyimino hydrogen to the C-2 of oxime (**1**).<sup>2</sup>

We have now carried out extensive investigations on the effects of the substitution by methyl groups of the hydrogens attached to the  $\alpha$ - or  $\beta$ -carbon of the  $\alpha,\beta$ -unsaturated hydroxyimino chromophore.

We now report that the effects of the alkyl group attached to the C-C double bond are profound and irradiation of (*E*)- $\beta$ -methyl- $\alpha,\beta$ -unsaturated oxime (**10**) in protic or aprotic solvent resulted in an unprecedented isomerization involving an intramolecular stereospecific transfer of the hydroxyimino proton.

The substrates (*E*)-4,4-dimethylcholest-1-en-3-one oxime (**3**) and (*E*)-1,4,4-trimethylcholest-1-en-3-one oxime (**10**) were synthesised as follows (Scheme 2); dimethylation of cholest-1-en-3-one (**5**) gave 4,4-dimethylcholest-1-en-3-one



**Scheme 2.** Reagents and conditions: i, MeI-Bu<sup>o</sup>OK-THF (tetrahydrofuran); ii, Me<sub>2</sub>CuLi-Et<sub>2</sub>O at 0 °C; iii, C<sub>5</sub>H<sub>5</sub>N·HBr-Br<sub>2</sub>-AcOH at 60 °C; iv, CaCO<sub>3</sub>-DMA (dimethylacetamide) reflux; v, NH<sub>2</sub>OH·HCl-AcONa-EtOH, reflux; vi, *hν*-MeOH or C<sub>6</sub>H<sub>6</sub> or C<sub>6</sub>D<sub>6</sub>; vii, *hν*-MeOD; viii, EtOD-Na.

(6)<sup>†</sup> (88%). Conjugate addition of a methyl group to 4,4-dimethylcholest-1-en-3-one (6) gave 1α,4,4-trimethyl-5α-cholestan-3-one (7)<sup>†</sup> (93%). Monobromination of trimethyl-

<sup>†</sup> (6), M.p. 90–91 °C. (7), M.p. 116–117 °C. (8), M.p. 48–50 °C.  $\nu_{max}$  1672 cm<sup>-1</sup> (C=O);  $\delta$  0.71 (3H, s, 18-H), 1.08 (6H, s, 4-Me<sub>2</sub>), 2.05 (3H, d, *J* 1.3 Hz, 1-Me) and 5.76 (1H, d, *J* 1.1 Hz, 2-H); *m/z* 426 (*M*<sup>+</sup>, 48.2).

(10), M.p. 180–181 °C;  $\nu_{max}$  3236 cm<sup>-1</sup> (OH);  $\delta$  0.69 (3H, s, 19-H), 1.14 (6H, s, 4-Me<sub>2</sub>), 2.02 (3H, d, *J* 1.1 Hz, 1-Me), 6.43 (1H, d, *J* 1.1 Hz, 2-H); *m/z* 441 (*M*<sup>+</sup>, 65.3) and 424 [(*M*-OH)<sup>+</sup>, 50.7].

(3), M.p. 196–197 °C. (4), M.p. 156–157 °C;  $\nu_{max}$  1620 cm<sup>-1</sup> (C=N);  $\delta$  0.64 (3H, s, 18-H), 0.90 (3H, s, 19-H), 1.16, 1.21 (each 3H, each s, 3-Me<sub>2</sub>), 3.44 (1H, dd, 4α'-H), 3.84 (1H, dd, 5β'-H) and 4.38 (1H, dd, 5α'-H).

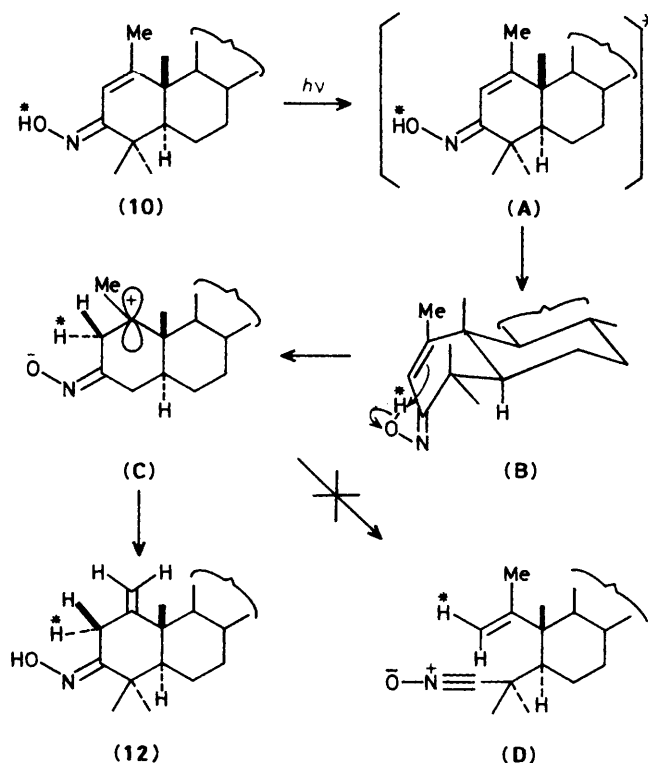
(12), M.p. 172–174 °C;  $\nu_{max}$  3262 cm<sup>-1</sup> (OH);  $\delta$  0.69 (3H, s, 18-H), 1.03 (3H, s, 19-H), 1.14 (6H, s, 4-Me<sub>2</sub>), 3.27 (1H, d, *J* 17.6 Hz, 2-H), 3.34 (1H, d, *J* 17.6 Hz, 2-H), 4.67 and 4.92 (each 1H, each br. s, -C-CH<sub>2</sub>); *m/z* 441 (*M*<sup>+</sup>, 100%), 426 [(*M*-Me)<sup>+</sup>, 89.1] and 424 [(*M*<sup>+</sup>-OH), 44.8].

(13), M.p. 169–172 °C;  $\nu_{max}$  3260 cm<sup>-1</sup> (OH);  $\delta$  0.69 (3H, s, 18-H), 1.03 (3H, s, 19-H), 1.14 (6H, s, 4-Me<sub>2</sub>), 3.39 (1H, s, 2β-H), 4.67 and 4.92 (each 1H, each s, -C-CH<sub>2</sub>); *m/z* 422 (*M*<sup>+</sup>, 50.7%) and 425 [(*M*-OH)<sup>+</sup>, 42.1].

(15), M.p. 184–187 °C;  $\nu_{max}$  3278 cm<sup>-1</sup> (OH);  $\delta$  0.69 (3H, s, 18-H), 1.01 (3H, s, 19-H) and 1.14 (6H, s, 4-Me<sub>2</sub>); *m/z* 445 (*M*<sup>+</sup>, 80.9%) and 428 [(*M*-OH)<sup>+</sup>, 53.7].

(14),  $\delta$  0.71 (3H, s, 18-H), 1.08 (6H, s, 4-Me<sub>2</sub>) and 1.10 (3H, s, 19-H); *m/z* 430 (*M*<sup>+</sup>, 39.8%).

(17), M.p. 160–163 °C;  $\nu_{max}$  3278 cm<sup>-1</sup> (OH);  $\delta$  0.69 (3H, s, 18-H), 1.03 (3H, s, 19-H), 1.13 (6H, s, 4-Me<sub>2</sub>) and 3.24 (1H, s, 2α-H); *m/z* 444 (*M*<sup>+</sup>, 67.6) and 427 [(*M*-OH)<sup>+</sup>, 68.5].



**Scheme 3**

5 $\alpha$ -cholestan-3-one (**7**) afforded 2 $\alpha$ -bromo-1 $\alpha$ ,4,4-trimethyl-5 $\alpha$ -cholestan-3-one (**8**)<sup>†</sup> (91%) which was dehydrobrominated to give 1,4,4-trimethyl-5 $\alpha$ -cholest-1-en-3-one (**9**)<sup>¶</sup> (91%).

Oximation of the enones (**9**) and (**6**) gave the oxime (**10**)<sup>†</sup> and (**3**).<sup>†</sup>

Irradiation of oxime (**3**) (118 mg) in methanol (60 ml) with a low pressure Hg arc in a Rayonet photochemical reactor for 5 h under nitrogen resulted in the formation of isoxazoline (**4**)<sup>†</sup> (24%) together with (*Z*)-4,4-dimethyl-1 $\beta$ -methoxy-5 $\alpha$ -cholestan-3-one oxime (23%), its (*E*)-isomer (12%) and the parent ketone (6%). These results almost parallel those obtained in the photolysis of 5 $\alpha$ -cholest-1-en-3-one oxime (**1**).<sup>2</sup>

In sharp contrast to these results, when oxime (**7**) (140 mg) in methanol (70 ml) was irradiated under the above conditions for 4 h, only the product (**12**) was obtained (Scheme 2). Neither the expected isoxazole derivative (**16**) nor the unsaturated lactam that arises from the photo-Beckmann rearrangement was found. An identical product was obtained when a benzene solution of oxime (**10**) was irradiated for 7 h (17%) or a hexadeuteriobenzene solution of oxime (**10**) was irradiated for 8 h (75%). The structure of the product (**12**) was assigned on the basis of the i.r., <sup>1</sup>H n.m.r., and mass spectra.<sup>†</sup>

The following deuterium labelling studies proved that the 2 $\alpha$ -H of photo-products (**12**) are stereospecifically derived from the hydroxyimino proton of oxime (**10**) and therefore the transfer is likely to be intramolecular.

Thus, when the hydroxyimino proton of oxime (**10**) is exchanged by deuterium by dissolution in MeOD and the resulting MeOD solution is irradiated under the above-mentioned conditions, a *single* monodeuteriated product (**13**)<sup>†</sup> is obtained. The <sup>1</sup>H n.m.r. spectrum exhibited a singlet at 3.39 assignable to 2-H. This indicated that the deuterium is stereospecifically incorporated into C-2. On the other hand, when tetradeuteriated oxime (**15**)<sup>†</sup>, prepared by oximation of the tetradeuteriated enone (**14**), obtained *via* deuteration of enone (**9**) with EtOD and Na, was photolysed in methanol,

the single trideuteriated photoproduct (**17**)<sup>†</sup> was obtained. The <sup>1</sup>H n.m.r. spectrum of product (**17**) exhibited a singlet at  $\delta$  3.24 assignable to 2-H. This again shows that the transfer of the proton is stereospecific.

The configurations of the incorporated 2-H in products (**13**) and (**17**) were assigned to be  $\beta$  and  $\alpha$  respectively on the basis of our previously proposed mechanism<sup>2</sup> and the assignments were independently confirmed by means of nuclear Overhauser enhancement (n.O.e.) measurements. Thus, while irradiation of the 19-H of product (**14**) resulted in the enhancement of the signal area of the 2-H, no enhancement of the 2-H was observed when the 19-H of product (**17**) was irradiated. Similarly, while irradiation of the 2-H of product (**14**) caused the enhancement of the signal area of the 19-H, no enhancement of the 19-H was observed when the 2-H of product (**17**) was irradiated.

On the basis of these labelling results, we consider that the path of the present photoinduced isomerization of oxime (**10**) into oxime (**12**) takes place as outlined in Scheme 3. Irradiation of the oxime (**10**) generates excited oxime (A) and the hydroxyimino proton is transferred to the C-2 of the twisted C=C bond of the relaxed unstable species (B) from the  $\alpha$ -face to give carbocation (C). A loss of a proton from the species (C) affords the isomeric oximes (**12**). In contrast to the photochemical behaviour of (*E*)-5 $\alpha$ -cholest-1-en-3-one oxime (**1**)<sup>2</sup> and the 4,4-dimethyl derivative (**3**), no product that might arise from a competing fragmentation of carbocation (C) to give a nitrile oxide (D) was formed in the present photo reaction.

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## References

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